

Second Semester BSc Degree examination may 2019

Foundation Course Chemistry

CH1221 – Methodology and Perspectives of Sciences and General Informatics

(2017 Admission onwards) Max Marks :80

Q.No	Section A Answer All questions, Each question carries 1 mark	Marks
1	Give the difference between science and technology Science – organized knowledge Technology – application of science	
2	What are the contributions of Dmitri Mendeleev Any one point related to periodic law / periodic table/ ----	
3	What is the SI unit of electric current Ampere	
4	Suggest an alternative for conventional fuel Solar energy/ wind energy/fuel cell Any one	
5	What is sampling Selection of a definite part of matter for analysis	
6	Who are the winners of Nobel prize in Chemistry in the year 2018 Frances H Arnold, George P Smith, Gregory P Winter	
7	What is plagiarism Deliberate representation of others words/ ideas and thoughts as one's own work without acknowledgement	
8	Define ionic product Product of molar concentrations of ions present in solution	
9	Name one structure drawing software in chemistry Chem pen/ chem. Sketch/ chem. Draw/ chem. Doodle.....	

10	Name one carcinogenic chemical used in the lab Give mark if attempted	
	Section B Answer any eight questions Each question carries 2 marks	
11	Distinguish between accuracy and precision Accuracy – concordance between experimental value and true value Precision – reproducibility / concordance of a series of measurements	
12	What is data sharing	
13	What is histogram	
14	What is correlation analysis Used to determine the amount of correlation between two or more variables Eg. $PV = nRT$, there is relation between P & T / or any other relation	
15	What is standard deviation	
16	What is virtual lab of MHRD Project of MHRD- to provide access to laboratories in various disciplines of science and engineering for students of all levels	2
17	What is meant by a primary standard in titrimetric analysis Definition / characteristics – any four points	2
18	Give two examples of redox indicators Diphenyl amine / N phenyl anthranilic acid /.....	
19	What is INFLIBNET	2
20	Explain the basic concepts of IPR Right of an individual/ organization for a creative work.	2

21	Name four toxic chemicals used in lab Any four	2
22	Write the smiles of cyclo hexane and benzene Cyclo hexane : C1CCCCC1 Benzene : c1ccccc1	2
Section C Answer any six questions, each question carries 4marks		
23	Write a note on combinatorial chemistry Computer simulations of different but structurally related molecules or materials -high impact in the pharmaceutical industry – virtual library of molecules-robotics Applications – developing catalysis- electronic devices	2 2
24	What is nano technology? Give two important applications of nano materials Definition – explanation Applications (any two)	2 2
25	Explain the term co precipitation with an example.	
26	What are the postulates of Bohr's atom model Five main postulates	4
27	Explain the theory of acid base titrations Explanation with examples – reactions- pH change near the end point – selection of indicators	4
28	Write four important principles of green chemistry Four principles	4
29	What is NPTEL Technology enhanced national learning programme- funded by MHRd-organised by IITs, IISc.....high quality learning materials ...	4
30	What are metallochromic indicators? Give examples. Definition and explanation	3

	example	1
31	<p>Explain copy right and patents</p> <p>Copy right – IPR provided by the government to authors of original work- literary-musical- computer programmes – other creative works -.....</p> <p>Patent – right to any one who invents or discovers any new and useful process/ machine / productownership of patent - procedure</p>	
	<p>Section D</p> <p>Answer any two questions, Each question carries 15marks</p>	
32	<p>Explain various types of file formats and databases in chem. Informatics</p> <p style="padding-left: 40px;">Basics of chem. Informatics – definition and need of chem. informatics</p> <p>Applications-</p> <p>various types of file formats-</p> <p>data bases</p>	<p>5</p> <p>5</p> <p>5</p>
33	<p>Write short note on</p> <p>(a) Common ion effect</p> <p>(b) Solubility product</p> <p>And explain their applications in inorganic qualitative analysis</p> <p>Common ion effect : definition / explanation with example</p> <p>Solubility product :definition / explanation with example</p> <p>Application in group analysis – with examples</p>	<p>5</p> <p>5</p> <p>5</p>
34	<p>Discuss the safety measures in the lab and mention the hazard warning information</p> <p>Safety measures-</p> <p>before starting work –</p> <p>during the lab-</p> <p>before leaving the lab-</p>	5x3=15

	<p>fire hazards-</p> <p>labels and material safety warnings</p>	
35	<p>Write short note on the following</p> <p>(a) Micro scale analysis</p> <p>(b) Redox titrations</p> <p>(c) TLC</p> <p>Micro scale analysis : green approach- reduce amount of chemicals- reduced amount of chemical hazards- less waste- reduce the release of toxicchemicals- economical – easier – reduce the time</p> <p>Redox titrations ; definition explanation with examples</p> <p>TLC – definition – procedure – explanation – rf value - advantages</p>	<p>5</p> <p>5</p> <p>5</p>